

Regioselective Introduction of 2-Propynyl Groups into the C-2 or C-3 Position of Furanoside Ring

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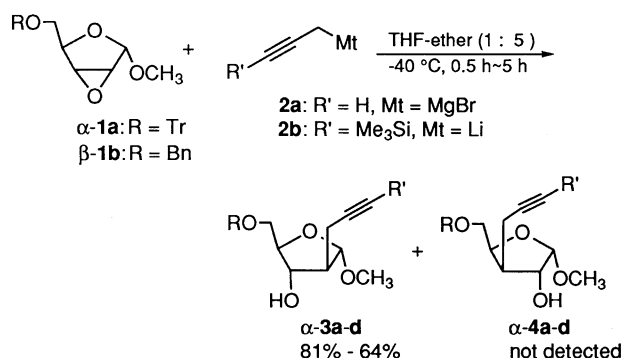
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The reaction of methyl 2,3-anhydro- α -D-ribofuranoside or methyl 2,3-anhydro-5,6-*O*-cyclohexylidene- α -D-allofuranoside with 2-propynyl metallic reagents exclusively gave the corresponding 2-*C*-(2-propynyl)-*arabino*-pentofuranosides and 2-*C*-(2-propynyl)-*altro*-hexofuranosides, respectively. In contrast, the β -anomers were selectively attacked at the position 3 to afford the corresponding 3-*C*-(2-propynyl)furanosides.

Carbohydrates have attracted attention as starting materials for the synthesis of natural products with multiple centers of chirality.¹ One of the most fundamental processes encountered in this approach is regioselective activation of the specific carbon atom of a carbohydrate. In this context, anhydropyranosides with locked conformation have been widely utilized in organic synthesis because the reaction site attacked by nucleophiles could be predicted in terms of diaxial oxirane-ring opening.² On the other hand, the reactions of 2,3-anhydrofuranosides with nucleophiles are more complicated, the reaction site being difficult to predict.³ In view of the clean regioselectivity observed in the reaction of α - and β -furanosides with allylic Grignard reagents,³ the same trend would be expected in the reaction of the furanosides with 2-propynyl metallic reagents. In this communication, we wish to report regioselective introduction of 2-propynyl groups into furanoside frameworks.

At the outset, methyl 2,3-anhydro-5-*O*-trityl- α -D-ribofuranoside (α -**1a**) was allowed to react with 2-propynyl-magnesium bromide (**2a**)⁴ in THF-ether (1 : 5) at -40 °C for 1.5 h to give methyl 2-deoxy-2-*C*-(2-propynyl)-5-*O*-trityl- α -D-*arabino*-pentofuranoside (α -**3a**) in 64% yield. Under similar conditions, the reaction of methyl 2,3-anhydro-5-*O*-benzyl- α -

D-ribofuranoside (α -**1b**) with **2a** gave the expected 2-*C*-(2-propynyl)-5-*O*-benzyl- α -D-*arabino*-pentofuranoside (α -**3b**) in 69% yield. In these reactions, no detectable formation of regioisomers α -**4a** and α -**4b** could be observed (Scheme 1; Table 1, entries 1 and 2).



Scheme 1. (For α -**3a-d** and α -**4a-d**, see Table 1).

3-Trimethylsilyl-2-propynyllithium (**2b**)⁵ reacted smoothly with α -**1a** and with α -**1b** to give the corresponding 2-deoxy-2-*C*-(3-trimethylsilyl-2-propynyl)-*arabino*-pentofuranosides α -**3c** and α -**3d** in 81% and 76% yields, respectively. In these reactions, no regioisomer α -**4c** or α -**4d** was again detected (Scheme 1; Table 1, entries 3 and 4).

Contrary to the cases of the α -anomers, the reactions of the corresponding β -anomers with either **2a** or **2b** were complicated. Thus, no reaction took place when β -**1a** was treated with **2a** in

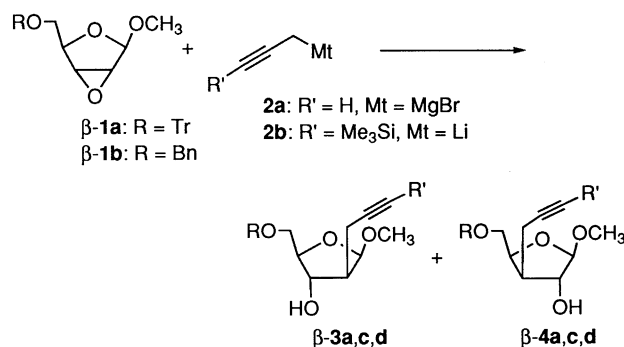
Table 1. Reaction of 2,3-anhydrofuranosides with 2-propynyl metallic reagents

Entry	Substituents of substrate, reagent, and product		Substrate	Reagent	Solvent ^a	Temp/°C	Time/h	Product: yield/% ^b		Recovery of substrate/%
1	Tr	H	α - 1a	2a	T-E	-40	1.5	α - 3a : 64	α - 4a : nd	nd
2	Bn	H	α - 1b	2a	T-E	-40	2.0	α - 3b : 69	α - 4b : nd	nd
3	Tr	TMS	α - 1a	2b	T-E	-40	0.5	α - 3c : 81	α - 4c : nd	nd
4	Bn	TMS	α - 1b	2b	T-E	-40	5.0	α - 3d : 76	α - 4d : nd	nd
5	Tr	H	β - 1a	2a	T	-10	24	β - 3a : nd	β - 4a : nd	90
6	Tr	H	β - 1a	2a	E	-10	24	β - 3a : nd	β - 4a : 36	63
7	Tr	H	β - 1a	2a	E	rt	24	β - 3a : nd	β - 4a : 99	nd
8	Tr	TMS	β - 1a	2b	E	-20	6	β - 3c : nd	β - 4c : nd	74
9	Bn	TMS	β - 1b	2b	E	-20	6	β - 3d : nd ^c	β - 4d : nd	nd
10	-	H	α - 5	2a	T-E	-40	2	α - 6a : 62	α - 7a : nd	nd
11	-	TMS	α - 5	2b	T-E	-40	0.25	α - 6b : 75	α - 7b : nd	nd
12	-	H	β - 5	2a	E	-10	24	β - 6a : nd	β - 7a : 89	nd
13	-	TMS	β - 5	2b	E	-20	6	β - 6b : nd	β - 7b : nd	14

^a T-E = THF : ether = 1 : 5. T = THF. E = ether. ^b nd = Could not be detected. ^c Although an unstable single product was obtained, the structure could not be determined.

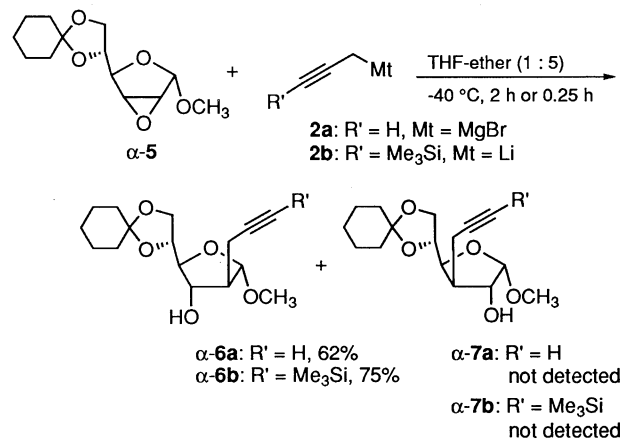
THF at -10 °C for 24 h (Table 1, entry 5), while, in ether at -10 °C, methyl 3-deoxy-3-*C*-(2-propynyl)- β -D-xylo-pentofuranoside (β -4a) was obtained in 36% yield with 63% recovery of β -1a (Scheme 2, Table 1, entry 6). When the reaction of β -1a with 2a was carried out in ether at room temperature for 24 h, β -4a was obtained in nearly quantitative yield (Scheme 2, Table 1, entry 7).

No reaction took place between 2b and β -1a (Scheme 2, Table 1, entry 8). On the other hand, 2b reacted with β -1b to afford a single product which could not be identified because of instability (Scheme 2, Table 1, entry 9).



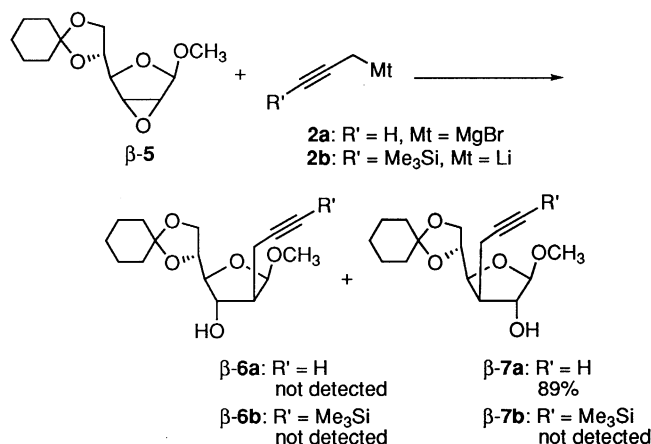
Scheme 2. (For β -3a,c,d and β -4a,c,d, see Table 1).

Reaction of methyl 2,3-anhydro-5,6-*O*-cyclohexylidene- α -D-allofuranoside (α -5) with 2a and with 2b in THF-ether (1 : 5, v/v) at -40 °C for 2 h gave methyl 5,6-*O*-cyclohexylidene-2-deoxy-2-*C*-(2-propynyl)- α -D-alto-hexofuranoside (α -6a) and methyl 5,6-*O*-cyclohexylidene-2-deoxy-2-*C*-(3-trimethylsilyl-2-propynyl)- α -D-alto-hexofuranoside (α -6b) in 62% and 75% yields, respectively, without any detectable formation of regioisomers α -7a and α -7b (Scheme 3, Table 1, entries 10 and 11).



Scheme 3.

Although β -5 reacted smoothly with the Grignard reagent 2a in ether at -10 °C to give β -7a in 89% yield, the reaction with the lithium reagent 2b resulted in the formation of a complex mixture of products (Scheme 4, Table 1, entries 12 and 13).



Scheme 4.

The observed regioselectivity may be rationalized on the bases of steric and electrostatic effects.³

The procedures described in this paper make 2-*C*-(2-propynyl)furanosides and 3-*C*-(2-propynyl)furanosides readily available. These compounds would be expected to have versatile utilities in synthetic organic chemistry.

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References and Notes

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- 3-Trimethylsilyl-2-propynyllithium (2b) was generated by the treatment of 1-trimethylsilyl-1-propyne with *tert*-butyllithium at -78 °C and then stirred at 0 °C for 1 h. a) E. J. Corey and C. Rücker, *Tetrahedron Lett.*, **23**, 719 (1982). b) K. Furuta, M. Ishiguro, R. Haruta, N. Ikeda, and H. Yamamoto, *Bull. Chem. Soc. Jpn.*, **57**, 2768 (1984).

This paper is dedicated to Professor Teruaki Mukaiyama on the occasion of his 70th birthday.